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Answer the following questions about the chemical equation
shown below $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ a What are the reactants'

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Directions Answer each of the following questions"***Chapter 7
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Balance the equations below
 $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$
 $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$
 $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$
 $4 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$

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ANSWERS 1 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 2 $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$ 3 $\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$ 4 $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ 5 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

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F₂ Cl₂ Br₂ I₂ The coefficients should add up to the number
at the end that is in parenthesis'**

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